

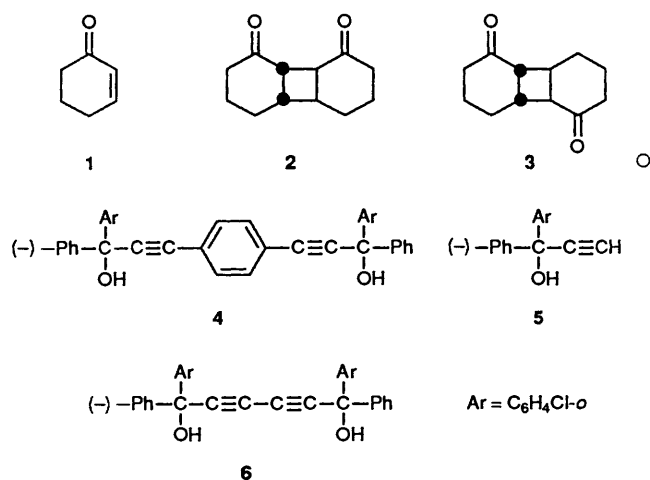
# Regio- and Enantio-selective Photodimerisation of Cyclohex-2-enone as an Inclusion Complex with a New Optically Active Host, (-)-1,4-Bis[3-(*o*-chlorophenyl)-3-hydroxy-3-phenylprop-1-ynyl]benzene: Preparation of the Optically Pure (-)-*syn-trans*-Dimer of Cyclohex-2-enone

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Photoirradiation of a 1:2 inclusion compound of (-)-1,4-bis[3-(*o*-chlorophenyl)-3-hydroxy-3-phenylprop-1-ynyl]benzene and cyclohex-2-enone as an aqueous suspension gave the (-)-*syn-trans*-dimer of cyclohex-2-enone in 48% ee, which on an optical resolution by complexation with (-)-1,6-bis(*o*-chlorophenyl)-1,6-diphenylhexa-2,4-diyne-1,6-diol afforded the optically pure enantiomer.

Photoirradiation of both neat and benzene solutions of cyclohex-2-enone **1** give a complex mixture of the *syn-trans*-**2** and *anti-trans*-dimer **3**,† and two other dimers of unknown structure.<sup>1</sup> Here, we report regio- and enantio-selective photodimerisation of **1** as an inclusion complex with a new (see later) chiral host, (-)-1,4-bis[3-(*o*-chlorophenyl)-3-hydroxy-3-phenylprop-1-ynyl]benzene **4** to give (-)-**2** in 48% ee selectively in 74.8% yield. Complexation of the dimer **2** (48% ee) with (-)-1,6-bis(*o*-chlorophenyl)-1,6-diphenylhexa-2,4-diyne-1,6-diol **6**,<sup>2</sup> afforded optically pure (-)-**2**. Photodimerisation of **1** proceeds more efficiently when the powdered inclusion complex of **1** and **4** is irradiated as an aqueous suspension rather than when it is irradiated directly.



A solution of **4** (5.0 g, 8.94 mmol) and **1** (1.72 g, 17.9 mmol) in ether-hexane (1:1, 10 cm<sup>3</sup>) was kept at room temperature for 6 h to give a 1:2 complex of **4** and **1** as colourless prisms (6.1 g, 91%), m.p. 90–95 °C. A suspension of the powdered complex (4.2 g) in a water (100 cm<sup>3</sup>) containing a small amount of sodium alkylsulfate as surfactant was irradiated‡ at room temperature for 24 h. The reaction mixture was filtered, dried, and distilled *in vacuo* to give (-)-**2** in 48.0% ee as an oil (0.8 g, 74.8%), [ $\alpha$ ]<sub>D</sub> -61.0§ (c 0.4, MeOH).

In contrast, irradiation of the solid powdered complex of **4** and **1** for 24 h, gave compound **2** in 46.5% ee (41.5% yield). It is

unclear why the photoreaction of the aqueous suspension is the more efficient process. The identity of **2** was established by comparison of its gas chromatographic retention time with that of an authentic sample prepared by a literature method.<sup>1</sup> The optical purity of **2** was determined by comparison of its [ $\alpha$ ]<sub>D</sub> value with that of an optically pure sample prepared as follows.

A solution of **6** (1.51 g, 3.13 mmol) and **2** (48.0% ee) (0.6 g, 3.13 mmol) in toluene (10 cm<sup>3</sup>) was kept at room temperature for 12 h to give a 1:1 complex of **6** and **2**. The complex was thrice recrystallised from toluene to give crystals of the pure complex; the latter when distilled *in vacuo* afforded optically pure **2** (0.27 g, 61.4%), [ $\alpha$ ]<sub>D</sub> -127 (c 0.56, MeOH). The optical purity of **2** was established on the basis that by further complexation with **6** its [ $\alpha$ ]<sub>D</sub> value was unchanged.

The mechanism of the regio- and enantio-selective photodimerisation of **1** in the inclusion complex with **4** will, it is hoped, be clarified by a future X-ray analysis of the complex.

For the selective photodimerisation of **1**, the new host **4** was the most effective whilst compound **6** was ineffective. The new host **4** was prepared by coupling an acetylenic compound and a halogenoarene.<sup>3</sup> Optically pure **5**<sup>4</sup> (53.5 g, 220 mmol), *p*-dibromobenzene (26.0 g, 110 mmol), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (0.1 g), CuI (0.1 g), PPh<sub>3</sub> (0.52 g) and Et<sub>3</sub>N (280 ml) were mixed and heated under reflux for 4 h. After remains of Et<sub>3</sub>N-HBr by filtration, the Et<sub>3</sub>N solution was evaporated to leave crude **4**. Recrystallisation of this from acetone gave a 1:2 complex of pure **4** and acetone as colourless prisms, m.p. 81–82 °C, which when heated *in vacuo* gave optically pure **4** as white powder [49.3 g, 80.2%, [ $\alpha$ ]<sub>D</sub> -62.6 (c 1.13, MeOH)], m.p. 82–83 °C.

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## References

- E. Y. Y. Lam, D. Valentine and G. S. Hammond, *J. Am. Chem. Soc.*, 1967, **89**, 3482.
- F. Toda, K. Tanaka, T. Omata, K. Nakamura and T. Ōshima, *J. Am. Chem. Soc.*, 1983, **105**, 5151.
- S. A. Jacobs and R. G. Harvey, *J. Org. Chem.*, 1983, **48**, 5134.
- F. Toda, K. Tanaka, H. Ueda and T. Ōshima, *Isr. J. Chem.*, 1985, **25**, 338.

† The structure was elucidated by X-ray crystal structure analysis. The data will be published in the near future.

‡ Photoirradiations were carried out using a 100 W Hg lamp.

§ [ $\alpha$ ]<sub>D</sub> Values are given in 10<sup>-1</sup> deg cm<sup>2</sup> g<sup>-1</sup> throughout.